Finite temperature quantum statistics of H₃⁺ molecular ion

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Full quantum statistical *NVT* simulation of the five-particle system H_3^+ has been carried out using the path integral Monte Carlo method. Structure and energetics are evaluated as a function of temperature up to the thermal dissociation limit. The weakly density dependent dissociation temperature is found to be around 4000 K. Contributions from the quantum dynamics and thermal motion are sorted out by comparing differences between simulations with quantum and classical nuclei. The essential role of the quantum description of the protons is established. © 2010 American Institute of Physics. [doi:10.1063/1.3464758]

I. INTRODUCTION

The triatomic molecular ion H_3^+ is a five-body system consisting of three protons and two electrons. Being the simplest polyatomic molecule, it has been the subject of a number of theoretical and experimental studies over the years.^{1–5} Experimentally, the H_3^+ ion was first detected in 1911 by Thompson;⁶ however, definite spectroscopic studies were carried out not until 1980 by Oka.⁷ Since then, this five-body system has proven to be relevant, also in astrophysical studies concerning the interstellar media and the atmosphere of gas planets. Therefore, low-density high-temperature H_3^+ ion containing atmospheres have been studied experimentally⁸ as well as computationally.⁹

Until now, the computational approaches have consistently aimed at finding ever more accurate potential energy surfaces (PESs) for H_3^+ at 0 K and consequent calculations of the rovibrational states.^{10,11} These calculations include Born–Oppenheimer (BO) electronic energies in various geometries often supplemented with adiabatic and relativistic corrections.^{12,13} For the study of rovibrational transitions, it is desirable to have an analytical expression for the PES, which is usually generated using Morse polynomial fits.¹⁰ Inclusion of the nonadiabatic effects, however, has turned out to be a cumbersome task, and so far they have not been rigorously taken into account.⁴

In this work, we evaluate the full five-body quantum statistics of the H_3^+ ion in a stationary state at temperatures below the thermal dissociation at about 4000 K. We use the path integral Monte Carlo (PIMC) approach, which allows us to include the Coulomb correlations between the particles exactly in a transparent way. Thus, we are able to monitor the fully nonadiabatic correlated quantum distributions of particles and related energies as a function of temperature. Furthermore, we are able to model the nuclei as classical mass points, in thermal motion or fixed as conventionally in quantum chemistry, and find the difference between these and the quantum delocalized nuclei.

The PIMC method is computationally expensive, but

within the chosen models and numerical approximations, it has been proven to be useful with exact correlations and finite temperature.^{14–21} For 0 K data with benchmark accuracies, however, the conventional quantum chemistry or other Monte Carlo methods, such as the diffusion Monte Carlo,²² are more appropriate. Thus, it should be emphasized that we do not aim at competing in precision or number of decimals with the other approaches. Instead, we will concentrate on physical phenomena behind the finite-temperature quantum statistics.

Next, we will briefly describe the basics of the PIMC method and the model we use for the ion. In Sec. IV, we first compare our 160 K PIMC "ground state" to the 0 K ground state and then consider the higher temperature effects.

II. METHOD

According to the Feynman formulation of the quantum statistical mechanics,²³ the partition function for interacting distinguishable particles is given by the trace of the density matrix

$$Z = \operatorname{Tr} \hat{\rho}(\beta) = \int dR_0 dR_1 \cdots dR_{M-1} \prod_{i=0}^{M-1} e^{-S(R_i, R_{i+1}; \tau)},$$

where $\hat{\rho}(\beta) = e^{-\beta \hat{H}}$, *S* is the action, $\beta = 1/k_B T$, $\tau = \beta/M$, $R_M = R_0$, and *M* is called the Trotter number. In this paper, we use the pair approximation in the action^{15,24} for the Coulomb interaction of charges. Sampling in the configuration space is carried out using the Metropolis procedure²⁵ with bisection moves.²⁶ The total energy is calculated using the virial estimator.²⁷

The error estimate in the PIMC scheme is commonly given in powers of the imaginary time time-step τ .¹⁵ Therefore, in order to systematically determine thermal effects on the system, we have carried out all the simulations with $\tau = 0.03 E_{\rm H}^{-1}$, where $E_{\rm H}$ denotes the unit of hartree. Thus, the temperatures and Trotter number *M* become fixed by the relation $T = (k_{\rm B}M\tau)^{-1}$.

In the following, we mainly use the atomic units, where the lengths, energies, and masses are given in units of the

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bohr radius (a_0) , hartree $(E_{\rm H})$, and free electron mass (m_e) , respectively. The statistical standard error of the mean (SEM) with 2SEM limits is used as an error estimate for the observables, unless otherwise mentioned.

III. MODELS

Two of the five particles composing the H_3^+ ion are electrons. For these, we do not need to sample the exact Fermion statistics, but it is sufficient to assign spin-up to one electron and spin-down to the other one. This is accurate enough, as long as the thermal energy is well below that of the lowest electronic triplet excitation.

We apply the same particle statistics for the three protons, too. This is even safer because the overlap of well localized nuclear wave functions is negligible and related effects become very hard to evaluate, anyway. On the other hand, however, the nuclear exchange due to the molecular rotation results in the so called zero-point rotations. These too contribute to energetics less than the statistical accuracy of our simulations. Therefore, we ignore the difference between ortho-H₃⁺ (*I*=3/2) and para-H₃⁺ (*I*=1/2). Thus, the protons are modeled as "boltzmannons" with the mass m_p =1.836 152 672 48×10³ m_e . The higher the temperature, the better is the Boltzmann statistics in describing the ensemble composed of ortho-H₃⁺ and para-H₃⁺.

For the *NVT* simulations, we place one H_3^+ ion into a cubic box with the volume of $(300a_0)^3$ and apply periodic boundary conditions and minimum image principle. This corresponds to the mass density of $\sim 1.255 \times 10^{-6}$ gcm⁻³. This has no essential effect at low *T*, but at high *T* the finite density gives rise to the molecular recombination balancing the possible dissociation. Within the considered temperature range, the contributions from the dissociated states are negligible.

The electrons are always simulated with the full quantum dynamics. For the nuclei, however, we use three models to trace the quantum and thermal fluctuations separately. The case of full quantum dynamics of all particles we denote by AQ (all-quantum), the mass point model of protons by CN (classical nuclei), and the adiabatic case of fixed nuclei by BO (Born–Oppenheimer potential energy surface).

IV. RESULTS AND DISCUSSION

A. Ground state: 0 K reference data

The equilibrium geometry of the H_3^+ ion in its ground state is an equilateral triangle D_{3h} for which the internuclear equilibrium distance is $R = 1.65a_0$.⁴ The best upper bound for the electronic ground state BO energy to date is $-1.343\ 835\ 625\ 02E_{\rm H}$.⁴ The vibrational normal modes of H_3^+ are the symmetric-stretch mode ν_1 and the doubly degenerate bending mode ν_2 . The latter one breaks the full symmetry of the molecule, and therefore it is infrared active.⁵

The vibrational zero-point energy is $0.019 \ 87E_{\rm H}$ and the so called rotational zero-point energies are $0.000 \ 29E_{\rm H}$ and $0.000 \ 40E_{\rm H}$ for para-H₃⁺ and ortho-H₃⁺, respectively.^{3,11} These yield about $0.020 \ 215E_{\rm H}$ for the average zero-point energy. Note, however, that the nuclear spins and zero point rotation are not included in our model of H₃⁺.



FIG. 1. Total energy of the H_3^+ molecular ion as a function of temperature. Fully nonadiabatic quantum statistical simulations [AQ (blue circles)], classical nuclei simulations [CN (red squares)], and the equilibrium geometry Born–Oppenheimer simulation [BO (black triangles)]. 0 K data (Refs. 3, 4, and 11) are given for comparison: BO ground state energy at equilibrium internuclear geometry (black dashed-dotted line), energy including the nuclear zero-point motion (green dashed line), and energy at the barrier to linearity (gray solid line). 2SEM statistical error estimate is shown by the error bars from simulations at the H_3^+ ion density ($300a_0$)⁻³ or ~1.255 × 10⁻⁶ g cm⁻³.

The lowest electronic excitation from the BO ground state is a direct Franck–Condon one $(0.710E_{\rm H})$ (Refs. 4 and 5) to dissociative potential curve: ${\rm H}_3^+ \rightarrow {\rm H}_2 + {\rm H}^+$ or ${\rm H}_3^+ \rightarrow {\rm H}_2^+$ +H.^{4,28} The dissociation energies (D_e) are $0.169E_{\rm H}$ and $0.241E_{\rm H}$, respectively.

The linear geometry with equal bond lengths $1.539\ 12a_0$ $(D_{\infty h})$ is a saddle point on the BO PES at $-1.278\ 681\ 90E_{\rm H}$ (Ref. 11) or 0.065 $15E_{\rm H}$ above the BO energy at the equilibrium geometry. This energy is usually called as the barrier to linearity.² The 0 K energetics is shown in Fig. 1 by the three horizontal lines.

B. PIMC ground state: 160 K

At our lowest simulation temperature $T \approx 160$ K, the electronic system is essentially in its ground state. For the total energy we find $-1.3438(2)E_{\rm H}$, see the BO black triangles in Fig. 1. The thermal energy is $k_{\rm B}T$ =0.000 507 $E_{\rm H}$, and therefore, the contribution from the rotational and vibrational excited states is also small and we find $-1.3406(29)E_{\rm H}$, see the CN red square in the same figure. The full quantum simulation includes vibrational zero-point contribution and yields $-1.3233(12)E_{\rm H}$, about $0.0205(14)E_{\rm H}$ above the BO energy in a good agreement with about $0.0202E_{\rm H}$ in Refs. 3 and 11.

From our AQ simulation we still find the equilateral triangle configuration of the nuclei with the internuclear distances increased to $\langle R \rangle = 1.723(4)a_0$, which indicates an increase of about $0.073(4)a_0$, as compared with the 0 K BO equilibrium distance bond lengths. Interestingly, within the error limits, this is the same as the bond length increase of the hydrogen molecule ion H₂⁺. The zero-point energy of H₃⁺ is about 2.7 times as large as that of the H₂⁺ ion,²¹ as expected from the increase of vibrational modes from one to three the zero-point energy of our model does not contain the rotational zero-point energy, as mentioned earlier.

The thermal motion (CN) alone increases the bond length to $\langle R \rangle = 1.658(4)a_0$ only (see the data in Figs. 2 and 3).



FIG. 2. Nuclear pair correlation functions (bond length distributions) at different temperatures from the quantum statistical simulations (solid lines) and from the classical nuclei simulations (dashed lines). The 0 K equilibrium internuclear distance is given as a vertical black dashed-dotted line. The distributions include the r^2 weight and normalization to unity. (Note that the r^2 weight is usually not included in description of extended or periodic systems.)

This clearly points out the difference between quantum and thermal delocalization of the nuclei at low T.

For the proton-electron and electron-electron interactions, the differences between our two approaches are smaller than in the proton-proton case but still distinctive. Comparison of the fixed nuclei simulation to the CN one shows that the two schemes give almost identical distributions. The AQ distributions, however, cannot be labeled identical with those from the CN or fixed nuclei simulations. The distributions are given in Figs. 4 and 5, where the notations are the same as in Fig. 2.

The calculations of the relativistic corrections involve, among other things, evaluation of the contact densities $\langle \delta(r_{ij}) \rangle$ for the electron-nuclei and the electron-electron pairs.¹² For the electron-nuclei contact density at the BO equilibrium configuration, we get 0.1814(20) and for the AQ case, 0.1765(20). For the electron-electron pair we get 0.0182(3) and 0.0166(3) for BO and AQ approaches, respectively. The estimated uncertainties due to extrapolation to the contact are given in parentheses. The 0 K reference values¹² for the BO case are 0.181 242 (electron-nuclei) and 0.018 386 63 (electron-electron). Thus, the quantum dynamics of the nuclei turns out to be a significant factor in lower-



FIG. 3. Expectation values of the internuclear distance at different temperatures from distributions in Fig. 2. Quantum statistical simulations (blue circles) and classical nuclei simulations (red squares). The FWHM limits are shown by triangles (all the lines are for guiding the eye). The 0 K equilibrium internuclear distance is shown as a horizontal black dashed-dotted line.



FIG. 4. Proton-electron pair correlation functions at the four temperatures from the full quantum statistical simulations [AQ (solid lines)] and from simulations with the classical nuclei [CN (dashed lines)]. That from the BO scheme is given at the lowest (electronic) temperature only (dashed-dotted line). Notations are the same as in Fig. 2.

ing the contact densities, too. See the snapshot of the AQ simulation in Fig. 6 for some intuition of the low-temperature quantum distributions in imaginary time.

C. High temperature phenomena

With the increasing temperature, the increasing contribution from rovibrational excitations is clearly seen in the total energies shown in Fig. 1. Contributions from the electronic excitations do not appear because the lowest excitation energy $0.710E_{\rm H}$ is much too high as compared to the thermal energy $k_{\rm B}T$. Consequently, the equilibrium geometry BO energy depends on the temperature almost negligibly. For convenience, the essential energetics related data have been collected into Table I also.

As expected, the increase in the total energy due to the classical rovibrational degrees of freedom is $9 \times \frac{1}{2}k_{\rm B}T$, defining the slope of the CN line. The most prominent quantum feature in AQ curve is, of course, the zero-point vibration energy. At higher temperatures, however, by comparing the AQ and CN curves we see that the quantum nature of nuclear dynamics becomes less important, except for dissociation.

At the dissociation limit we find the molecule with quantum nuclei somewhat more stable than the one with classical nuclei. With the relatively low density $(300a_0)^{-3}$, the molecule is mainly kept in one piece above 4000 K in the former case, whereas more dissociated in the latter. The total energy becomes higher for the CN than the AQ case slightly below



FIG. 5. Electron-electron pair correlation functions from the same simulations as those in Fig. 4. Notations are the same as in Figs. 2 and 4.



FIG. 6. xy-plane (z-projection) snapshot of the H₃⁺ ion from quantum statistical simulation with Trotter number 216, i.e., temperature of about 160 K, for all particles. "Polymer rings" describing the electrons are in the background (green and blue) and those of the nuclei are placed on top (yellow).

4000 K (see Table I). The total energies at this crossing point are above the "barrier to linearity,"^{2,11} already.

At higher temperatures $T \ge 4100$ K, other configurations, such as H_2+H^+ , H_2^++H , and $2H+H^+$, start playing more significant roles in the equilibrium dissociationrecombination processes. These will be considered in our next study.

The nuclear pair correlation function or bond length distributions (Figs. 2 and 3) follow the energetics discussed above. There, the zero-point vibration in AQ case is seen even better. At the 0 K limit both the expectation value and the distribution, in particular, are significantly different from those of the CN case.

The temperature dependence in the other pair correlation functions is weak (see Figs. 4 and 5). Obviously, this is the case, because electrons do not present a quantum-to-classical transition in the temperature range considered, now. Thus, the evolution in distributions in Figs. 4 and 5 following the rising temperature arises from the changes in the nuclear dynamics, and mostly, from the change in the conformation or the bond lengths, presented in Fig. 3.

V. CONCLUSIONS

In this study, the path integral Monte Carlo method was shown to be a successful approach for examination of quantum statistics of the five-particle molecule, H_3^+ ion. The method is based on the finite temperature mixed state de-

TABLE I. Energetics of the H₃⁺ molecular ion. The energies are given in units of hartree (atomic units). Simulation data are given with 2SEM error estimates. BO refers to Born-Oppenheimer calculation at equilibrium geometry. The reference data are rounded to convenient accuracy. The barrier to linearity is 0.065 $15E_{\rm H} \approx 1.8\,$ eV above the $E_{\rm BO}$ at 0 K.

	<i>T</i> (K)	$E_{\rm BO}$	$E_{\rm CN}$	$E_{\rm AQ}$
Reference 4	0	-1.343 836		-1.323 568 ^a
PIMC	~161	-1.3438(2)	-1.3406(29)	-1.3233(12)
PIMC	$\sim \! 1365$		-1.3236(8)	-1.3142(4)
PIMC	~ 2570		-1.3033(7)	-1.2977(6)
PIMC	~3856	-1.3438(2)	-1.2810(8)	-1.2770(2)
PIMC	~3999		-1.1469(9)	-1.2750(4)
PIMC	$\sim \! 4050$			-1.2734(9)

^aFor ortho-H₃⁺ estimated by using Refs. 3 and 4.

scription, and thus, it gives information, which is complementary to the high-accuracy 0 K description of conventional quantum chemistry. It was also shown how contributions from quantum and thermal dynamics to particle distributions and correlation functions can be sorted out, and furthermore, quantum-to-classical dynamics transition can be monitored.

Our approach is fully basis set and trial wave function free. It is based on the Coulomb interactions only and allows the most transparent interpretation of consequent particleparticle correlations.

Simulation at 160 K essentially reproduces the 0 K data from conventional quantum chemistry. Of course, a proper extrapolation to 0 K can be done for more accuracy. BO potential energy surface and the equilibrium geometry can be found by using classical nuclei with fixed coordinates. Description of the zero-point motion within our nonadiabatic five-body quantum simulation gives the vibrational zeropoint energy accurately. We find an increase of $0.073(4)a_0$ in the bond length due to the nonadiabatic zero-point vibration. The classical thermal contribution at 160 K is $0.008(4)a_0$, only.

With the raising temperature the rovibrational excitations contribute to the energetics, as expected, whereas the electronic part remains in its ground state in the spirit of BO approximation. At about 4000 K the H_3^+ ion dissociates, weakly depending on the ion density. We find that the full quantum molecule dissociates at slightly higher temperature compared to the one where the nuclei are modeled by classical particles with thermal dynamics only. Thus, we conclude the necessity of the quantum character of the protons in the correct description of dissociation.

We find that the nuclear quantum dynamics has a distinctive effect on the pair correlation functions, too. This is least for the electron-electron pair correlation function, stronger for the electron-proton one and largely increased in the proton-proton correlations. These are seen in the contact densities and, consequently, in the relativistic corrections where relevant.

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